

## 2,4-D and the environment

2,4-D has a relatively short half-life and is rather immobile in the soil. In 35 recent field dissipation studies across the United States, less than 5 per cent of applied 2,4-D moved downward more than 15 cm (6 inches). The average lowest depth recorded ranged from 6 to 12 inches in soils of the southern U.S. and 16 to 24 inches in low organic soils where more movement is expected. Soils were sampled to a depth of 48 inches and analyzed for 2,4-D plus its soil metabolites until two analyses provided a result of “non-detectable” at each sampled depth. Even though laboratory solubility studies indicated 2,4-D is potentially mobile, rapid degradation in the soil and removal from soil by plant uptake minimizes leaching under typical application conditions.

Field dissipation studies found that 2,4-D had an apparent soil half-life of five days with a range of 1.7 to 13.1 days. The moisture content of the soil appears to have a major effect on the half-life, since the main route of degradation is by microorganisms.

The commonly used 2,4-D amine salts and 2,4-D esters are not persistent under most environmental conditions. Dissociation of 2,4-D amine salt is expected to be instantaneous (<3 minutes) under most environmental conditions. Ester forms of 2,4-D biotransform and hydrolyze rapidly to the acid in natural soil (1 to 10 days) and natural water conditions (<1 day). Under these conditions, the environmental exposure from 2,4-D esters and 2,4-D amines is expected to be minimal in both terrestrial and aquatic environments.

The average half-lives of 2,4-D in grass and thatch were < 7 days. The half-life in natural water was one to two weeks, although in areas such as a treated rice paddy, the half-life was as short as one day.

The acid form of 2,4-D, as well as the amine and ester chemical groups, metabolize to transitory compounds of non-toxicological significance. Thus, 2,4-D is considered a biodegradable compound. Under normal conditions, 2,4-D residues are not persistent in soil, water, or vegetation.

For the most recent published study, which reviews the 35 environmental fate studies funded by the Task Force II, plus other 2,4-D environmental fate studies in scientific literature, (see Wilson, R.D et al, Dissipation in Field Soils after Application of 2,4-D Dimethylamine Salt and 2,4-D 2-Ethylhexyl Ester, *Environmental Toxicology and Chemistry*, Vol. 16(6), pp. 1239-1246, 1997), copies of which can be made available through this task force, or seen on the 2,4-D review publications site.

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## About the Task Force

The Industry Task Force II on 2,4-D Research Data is organized to provide funding for some 300 Good Laboratory Practice (GLP) research studies required to respond to the EPA re-registration and PMRA pesticide re-evaluation programs. The 2,4-D Task Force is comprised of those companies owning the technical registrations on the active ingredient in 2,4-D herbicides. They are Dow AgroSciences (U.S.), Nufarm, Ltd. (Australia) and Agro-Gor Corp., a U.S. corporation jointly owned by Albaugh, Inc. (U.S.) and PBI Gordon Corp. (U.S.).

## Environmental Fate Values

Adsorption/Desorption (Koc): 31.2 - 470.9 mL/g  
 Soil DT50 aerobic (20°C laboratory): 1.7 days  
 Soil DT50 anaerobic (20°C laboratory): 333 days (anaerobic aquatic study)  
 Hydrolysis (pH 7): 1 to 2 years in buffer solutions (sterile water)  
 Photolysis water DT50: 12.98 days in artificial sunlight  
 Photolysis soil DT50: 68 days  
 Aerobic Soil T1/2: 6.2 days  
 Aerobic Aquatic T1/2: 15 days  
 Anaerobic Aquatic T1/2: 41-333 days  
 Vapor Pressure:  $1.4 \times 10^{-7}$  mm Hg at 25°C (non-volatile)

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